

Díaz-Galiano, F. J.; Murcia-Morales, M.; Gómez-Ramos, M. M.; Fernández-Alba, A. R.

AgriFood Campus of International Excellence ceiA3 (ceiA3), Department of Chemistry and Physics, University of Almería e-mail: diaz-galiano@ual.es

Overview

Anthraquinone (9,10-dioxoanthracene) is an aromatic organic compound of risk to human health and to the environment. It can be found in food due to its formation and adsorption during **combustion processes**, which are used to **dry vegetables** like tea, camomile or coffee. The European Commission recommends its control at low concentrations, since these vegetables' consumption is high [1].



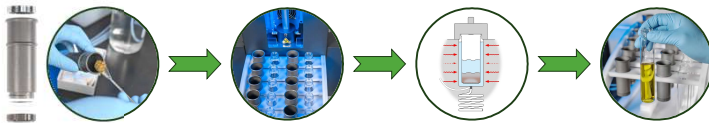
The **objectives** of this study are (i) to evaluate the difficulties associated with the currently used **multiresidue methods** for food within the European Union; (ii) to develop an automated extraction method that allows the control of anthraquinone at very **low concentrations** –below the current maximum residue limit (MRL) of 20 µg/kg for tea, coffee, herbal infusions and cocoa [2]; and (iii) to perform a **sampling study** of potentially contaminated dry vegetables.

Methods

Table 1. Automated pressurized liquid extraction method development parameters using the EDGE instrument.

Method (AMXX)	Solvent	Volume (mL)	Bubbling time (s)	Hold time (s)	T (°C)	Rinse step	Rinse volume (mL)	Total solvent (mL)	Dilution factor (V/m)	Clean-up (dSPE)	LOQ (µg/kg)	Rec. (%) (n = 3)
AM01	AcN	10	-	120	40	No	-	10	2.50	-	20	46
AM02	AcN	10	-	120	40	No	-	10	2.50	PSA	20	49
AM03	AcN	10	-	120	40	No	-	10	2.50	PSA, FA	20	54
AM04	AcOEt	10	-	120	40	No	-	10	2.50	-	20	27
AM05	AcOEt	10	-	120	40	No	-	10	2.50	PSA	20	29
AM06	AcOEt	10	-	120	40	No	-	10	2.50	PSA, FA	20	28
AM07	AcN	10	60	60	40	No	-	10	2.50	-	> 20	-
AM08	AcN	10	90	60	40	No	-	10	2.50	-	20	31
AM09	AcN	5	60	60	40	Yes	5	10	2.50	-	20	41
AM10	AcN	10	-	90	40	Yes	5	15	3.75	-	10	78
AM11	AcN	10	30	90	40	Yes	5	15	3.75	-	20	57
AM12	AcN	10	-	150	40	Yes	5	15	3.75	-	7.5	101

AcN = acetonitrile; AcOEt = ethyl acetate; FA = formic acid; PSA = primary secondary amine




Automated sample extraction with pressurized liquid extraction was performed with the aid of an EDGE instrument. The amount of Earl Grey tea used in every method was 4 g. The limit of quantitation (LOQ) was determined by spiking three tea aliquots with anthraquinone (AQ) at three concentration levels (7.5, 10 and 20 µg/kg) (**Table 1**).

Deuterated anthraquinone (AQ-D₈) was employed as a means to correct any matrix effects that could arise from the studied methods, and to evaluate its use as a surrogate standard for AQ. The spiked levels of AQ-D₈ were also 7.5, 10 and 20 µg/kg.

All analyses were carried out using gas chromatography coupled to tandem mass spectrometry (GC-QqQ-MS/MS) (**Figure 1**).

Sample extraction
EDGE automated extraction system



GC parameters
Agilent Intuvo 9000 system
2 planar columns
(15 m × 0.25 mm × 0.25 µm)
Splitless injection
Total runtime 12.4 min
Injection volume 1 µL

MS parameters
Agilent 7010B Triple Quad
Ionization mode: EI
Acquisition mode: dMRM

AQ transitions:
208-152-180-152-208-180
AQ-D₈ transitions:
216-160-188-160-216-188




Figure 1. Sample extraction and GC-QqQ-MS/MS analysis.

Results

An already established multiresidue method including a sample hydration step [3] was tested, providing poor results. With the hypothesis that sample hydration was causing the coextraction of interfering matrix compounds in the analysis of AQ, automated pressurized liquid extraction was studied (**Table 1**).

First, AcN was determined to be the most appropriate extraction solvent, as it provided higher recovery values than AcOEt. Moreover, the use of a clean-up step including PSA was found to be unnecessary (**AM01-AM06**). A bubbling step to increase sample and solvent mixing was tested, but it was deemed counterproductive (**AM07-AM09**). Furthermore, a rinse step was also evaluated, and while it increased recovery values, it also meant a greater dilution factor. In the end, increasing the hold time (time the sample within the solvent) and using a rinse step were the crucial steps in the obtention of a low LOQ (7.5 µg/kg) and a 101 % recovery value (**AM12**).

Multiresidue method [3]

2 g tea + 4 mL H₂O (wait 30 min) + 10 mL AcN + 200 µL IS (25 mg/L)

Shake 7 min

4 g MgSO₄, 1 g NaCl, 1 g Na₂Citrate-2 H₂O, 0.5 g Na₂H₂Citrate-1.5 H₂O

Shake 7 min Centrifuge 5 min

dSPE: 3 mL extract + 150 mg CaCl₂ + 150 mg PSA

Vortex 30 s Centrifuge 5 min

Transfer supernatant into 4 mL vial + add 30 µL formic acid (5 %)

Evaporate 50 µL of the extract and dissolve in AcOEt

ANALYSIS

Method AM12

4 g tea

Cover with sand or Q-Screen

10 mL AcN

Hold 150 s at 40 °C

5 mL AcN (rinse)

Evaporate 50 µL of the extract and dissolve in AcOEt

ANALYSIS

SAMPLE THROUGHPUT
AM12: ~70/8 hr
Manual: ~30/8 hr

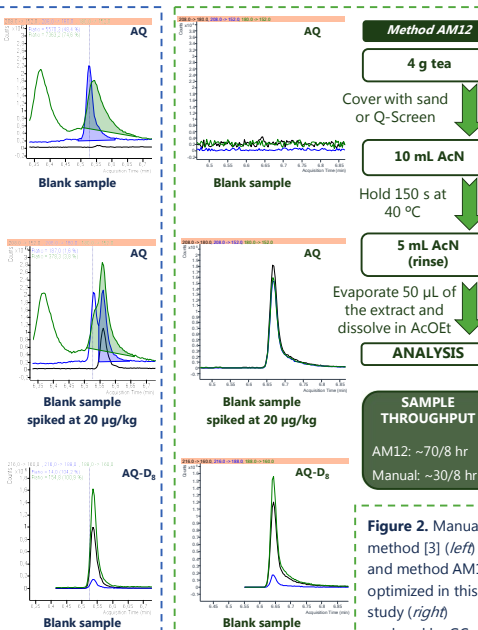


Figure 2. Manual method [3] (left) and method AM12, optimized in this study (right) analysed by GC-QqQ-MS/MS.

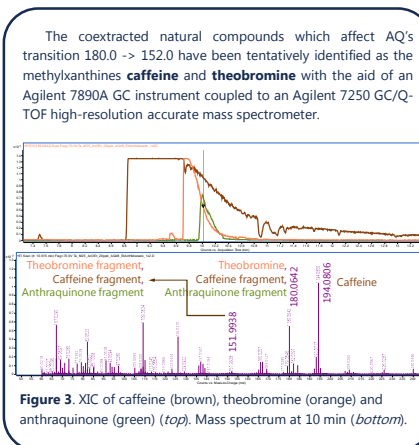
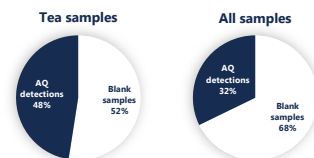


Figure 3. XIC of caffeine (brown), theobromine (orange) and anthraquinone (green) (top). Mass spectrum at 10 min (bottom).

The developed automated method has also been validated for **cocoa, coffee** and other **dry herbs**, at the same LOQ.



A sampling study was undertaken using **AM12**, during which AQ was detected in 32 % of the analysed samples (48 % if only tea samples are considered) at levels between the LOQ and the MRL.

Conclusions

- Using an automated extraction method based on **pressurized liquid extraction** avoids the coextraction of matrix interferences since it **does not require a hydration step**.
- Caffeine** and **theobromine** have been tentatively identified as the **main coextracted interferences** in manual extraction methods with a hydration step.
- The new **limit of quantitation is 7.5 µg/kg**, significantly lower than the current maximum residue level established by the European Commission.
- A sampling study of dry vegetables has detected **anthraquinone in 32 % of the samples** at levels **below the current maximum residue limit**, showcasing its better performance compared to manual extraction methods.
- Automated sample extraction instruments—including those based on pressurized liquid extraction—represent a very promising step forward in sample treatment and analysis, and can offer a **higher sample throughput** than manual extraction methods.